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ET and TTF Salts with Lanthanide Complex Ions Showing Frequency-Dependent ac Magnetic Susceptibility

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We measured conducting and magnetic properties of $(ET)_4[Ln(NCS)_6]\cdot CH_2Cl_2$ ($Ln = Dy, Ho, Er$, and Yb), $(ET)_5[Ln(NCS)_6NO_3]\cdot EtOH$ ($Ln = Tb$ and Dy), and $(TTF)_6[Dy(NCS)_6]$. All of the complexes behaved as semiconductors irrespective of Ln ions used, and also exhibited anomalous frequency-dependence of ac magnetic susceptibility below ca. 10 K, especially for Dy derivatives, indicating blocking behavior as found in single-molecule magnets.

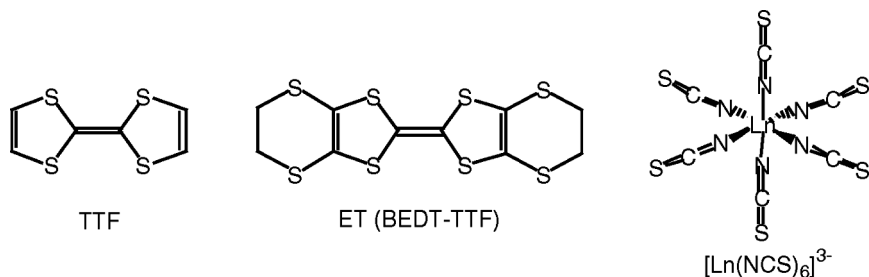
Keywords: BEDT-TTF; lanthanide ions; lanthanoid ions; magnetic properties; organic semiconductors; single-molecule magnets

INTRODUCTION

Search for suitable counter-ions is one of the main strategic tasks in the preparation of multi-functional organic conductors and superconductors, based on the salts of TTF (tetrathiafulvalene) and ET

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SCHEME 1 Structural formulas.

(BEDT-TTF; bis(ethylenedithio)TTF) (Scheme 1). We have already prepared several lanthanide (Ln) complexes showing frequency-dependent ac magnetic susceptibility [1], which can be observed usually for single-molecule magnets (SMMs) [2]. We then turned our attention to study on conducting SMMs obtained from TTF and ET with lanthanide complex ions.

EXPERIMENTAL

There have been pioneering works on ion-radical salts containing [Ln(NCS)₆]³⁻ [3] and [Ln(NCS)₆NO₃]⁴⁻ [4]. The present materials, (ET)₄[Ln(NCS)₆]·CH₂Cl₂ (abbreviated as **1Ln**; Ln = Dy, Ho, Er, and Yb) and (ET)₅[Ln(NCS)₆NO₃]·EtOH (**2Ln**; Ln = Tb and Dy), were prepared by means of the electro-crystallization method. We also prepared (TTF)₆[Dy(NCS)₆] (**3Dy**) by a chemical process using (TTF)₃(BF₄)₂. The composition was estimated by elemental analysis, IR spectrometry, and/or X-ray crystallographic analysis.

X-ray diffraction data were collected on a Rigaku R-axis RAPID diffractometer with graphite monochromated Mo K α radiation. The structures were solved by direct methods and expanded using Fourier techniques in the CrystalStructure program package [5]. Selected crystal data for **2Tb**: monoclinic, $P2_1/n$, $a = 17.76(1)$, $b = 15.57(1)$, $c = 33.00(2)$ Å, $\beta = 98.09(4)^\circ$, $Z = 4$, $V = 9033(8)$ Å³, $d_{\text{calc}} = 1.867$ g cm⁻³, $T = 90$ K.

Conductivity and magnetic measurements were performed on a Quantum Design PPMS. Single-crystal conductivities were measured along a long axis of the platelet-shaped crystals of **1Ln** and **2Ln**. Ac magnetic susceptibilities of randomly oriented polycrystalline samples were measured varying the frequency (50–10000 Hz) of an applied ac magnetic field (amplitude: 5 Oe).

RESULTS AND DISCUSSION

The crystal structures of **1Ln** (Ln = Ho, Er, Yb) [3] and **2Dy** [4] are described previously. New complex **2Tb** was isomorphous to **2Dy** [4]. The molecular arrangement in a donor sheet of **2Tb** is shown in Figure 1(a). Salts **3Ln** did not afford good crystals suitable for single-crystal X-ray diffraction study.

The room-temperature resistivity of **1Ln** was ca. $20\ \Omega\text{cm}$ with the activation energy (E_a) of 0.08–0.4 eV. The resistivities of **2Ln** were ca. $10\ \Omega\text{cm}$ with $E_a = 0.09$ and 0.1 eV for Ln = Tb and Dy, respectively (Fig. 1(b) for **2Tb**). Compressed pellets of **3Dy** were also semiconducting with $E_a = 0.08$ eV. All of the ion-radical salts investigated here are semiconductors. Tamura *et al.* reported metallic conductors using BO [6] and TTP [7] with $[\text{Ln}(\text{NCS})_6]^{3-}$ and $[\text{LnCl}_4(\text{H}_2\text{O})_6]^-$ [8]. The choice of donor molecules is crucial for development of metallic conductors especially when bulky counter anions are incorporated.

Ac magnetic susceptibilities were measured for polycrystalline samples of **1Ln** and **2Ln** and fine powder samples of **3Dy**. Frequency dependence of χ''_{ac} was observed for **1Ln**, where χ'_{ac} and χ''_{ac} denote the in-phase and out-of-phase parts of ac magnetic susceptibility, respectively. As shown in Figure 2 for example, χ''_{ac} of **1Dy** and **1Er** increased with an increase of frequency in a low temperature region (≤ 10 K). Very slight frequency dependence was found for both **2Tb** and **2Dy** (not shown). Complex **3Dy** showed an appreciable increase of χ''_{ac} with an increase of frequency (Fig. 3). The coordination geometry around

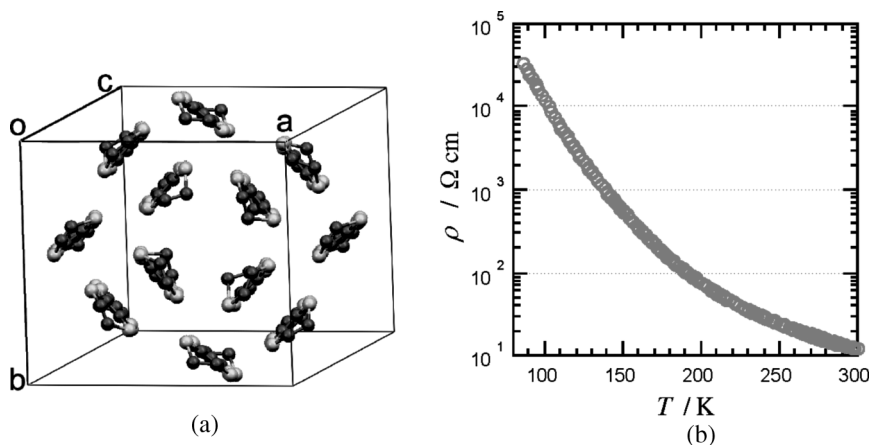


FIGURE 1 (a) Packing of ET molecules in the crystal of $(\text{ET})_5[\text{Tb}(\text{NCS})_6\text{NO}_3] \cdot \text{EtOH}$ (**2Tb**) viewed along the molecular axis. (b) Resistivity of **2Tb** measured along the ab diagonal (a - b) direction.

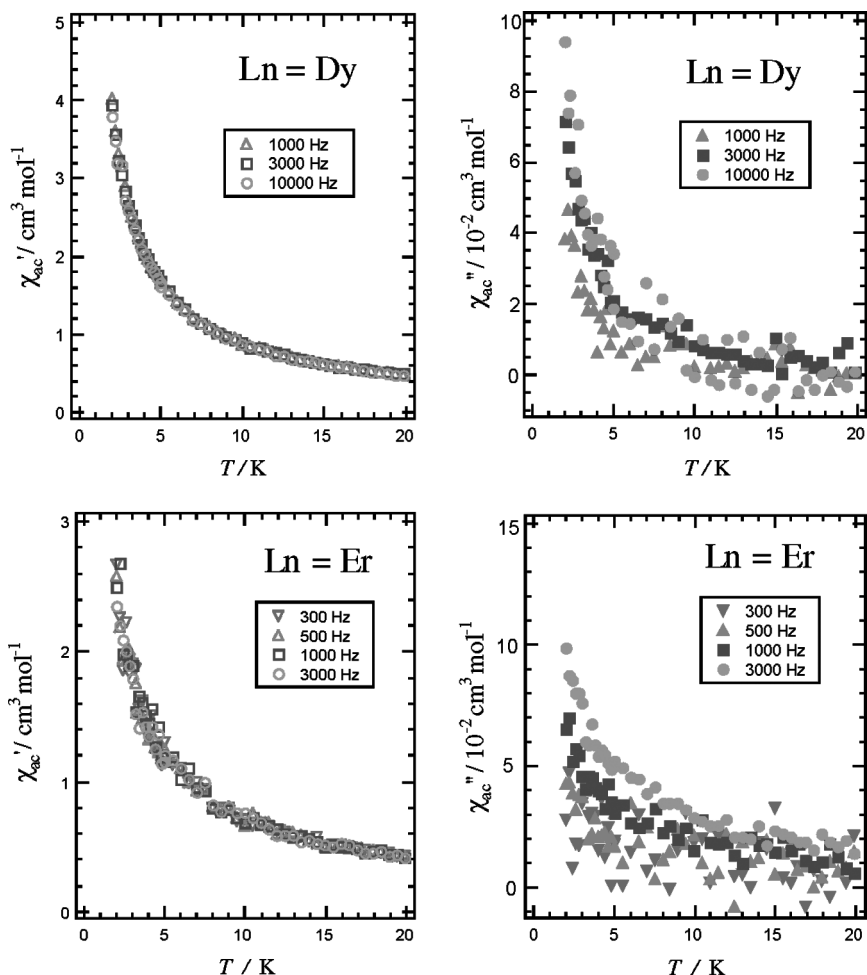


FIGURE 2 Frequency-dependence of χ'_{ac} and χ''_{ac} for $(\text{ET})_4[\text{Ln}(\text{NCS})_6] \cdot \text{CH}_2\text{Cl}_2$ (**1Ln**; Ln = Dy, Er).

the Ln ion in $[\text{Ln}(\text{NCS})_6]^{3-}$ is a highly symmetrical octahedron but that of $[\text{Ln}(\text{NCS})_6\text{NO}_3]^{4-}$ is unsymmetrical octa-coordinate due to additional oxygen atoms of NO_3 . Irrespective of the symmetries around the Ln ions, all of **1Ln–3Ln** exhibited relatively small frequency dependence of ac magnetic susceptibility. In contrast to our expectation, **1Ln** exhibited larger frequency dependence than **2Ln**.

These frequency dependences were substantial, and can be regarded as an indication of blocking behavior observable for SMMs

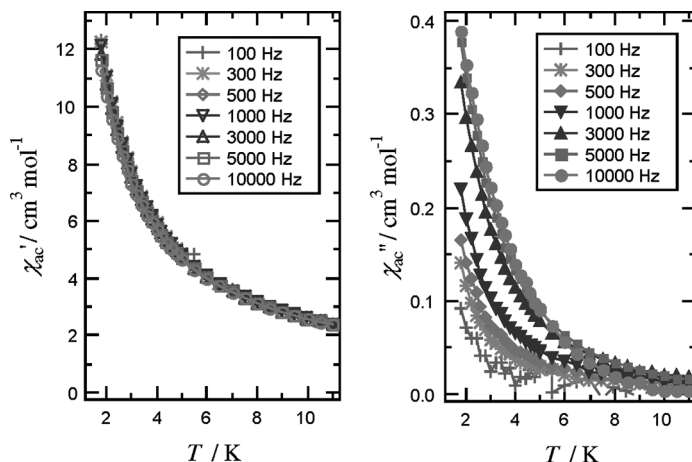


FIGURE 3 Frequency-dependence of χ'_{ac} and χ''_{ac} for $(\text{TTF})_6[\text{Dy}(\text{NCS})_6]$ (**3Dy**). Solid lines are shown for a guide to the eye.

due to slow magnetization relaxation [2]. These complexes might be semiconductors with magnetization blocking, but the blocking will take place at temperatures much lower than 2 K. To raise the possible blocking temperature, we have to turn our attention to other types of Ln-based counter anions with well-designed ligands. The $[\text{Ho}(\text{pc})_2]^-$ complex anion behaving as a SMM ($\text{pc} = \text{phthalocyaninate}^{2-}$) [9] may afford a clue to development of this study.

REFERENCES

- [1] Ueki, S., Sahlan, M., Ishida, T., & Nogami, T. (2005). *Synth. Met.*, *154*, 217; Mori, F., Ishida, T., & Nogami, T. (2005). *Polyhedron*, *24*, 2588; Mori, F., Nyui, T., Ishida, T., Nogami, T., Choi, K.-Y., & Nojiri, H. (2006). *J. Am. Chem. Soc.*, *128*, 1440.
- [2] Gatteschi, D., Caneschi, A., Pardi, L., & Sessoli, R. (1994). *Science*, *265*, 1054; Christou, G., Gatteschi, D., Hendrickson, D. N., & Sessoli, R. (2000). *MRS Bull.*, *25*, 66.
- [3] Tamura, M., Matsuzaki, F., Nishio, Y., Kajita, K., Kitazawa, T., Mori, H., & Tanaka, S. (1999). *Synth. Met.*, *102*, 1716.
- [4] Kazheva, O. N., Gener, M., Gritsenko, V. V., Kushch, N. D., Canadell, E., & Dyachenko, O. A. (2001). *Mendeleev Commun.*, *11*, 182.
- [5] CrystalStructure, version 3.5.1, Rigaku/MSO, The Woodlands, TX, USA, 2003.
- [6] Yamochi, H., Horiuchi, S., Saito, G., Kusunoki, M., Sakaguchi, K., Kikuchi, T., & Sato, S. (1993). *Synth. Met.*, *56*, 2096.
- [7] Misaki, Y., Fujiwara, H., Yamabe, T., Mori, T., Mori, H., & Tanaka, S. (1994). *Chem. Lett.*, 1653.

- [8] Tamura, M., Yamanaka, K., Mori, Y., Nishio, Y., Kajita, K., Mori, H., Tanaka, S., Yamaura, J.-I., Imakubo, T., Kato, R., Misaki, Y., & Tanaka, K. (2001). *Synth. Met.*, 120, 1041; Tamura, M., Imakubo, T., Yamanaka, K., Mori, Y., Nishio, Y., Kajita, K., Misaki, Y., Tanaka, K., Mori, H., Tanaka, S., Yamaura, J.-I., & Kato, R. (2002). *Mol. Cryst. Liq. Cryst.*, 379, 41.
- [9] Ishikawa, N., Sugita, M., & Wernsdorfer, W. (2005). *J. Am. Chem. Soc.*, 127, 3650.